

# The first in-situ observations of stratospheric $N_2O_5$ constrain heterogeneous chemistry on stratospheric aerosol

Zachary C.J. Decker<sup>1,2</sup>, Gordon A. Novak<sup>2</sup>, Maya Abou-Ghanem<sup>1,2</sup>, Steven S. Brown<sup>2</sup>, Paul T. Bui<sup>3</sup>, Glenn S. Diskin<sup>3</sup>, Jonathan Dean-Day<sup>4</sup>, Colin Gurganus<sup>1,2</sup>, Chris Jernigan<sup>1,2</sup>, Michael Lawler<sup>1,2</sup>, Daniel Murphy<sup>2</sup>, Michael A. Robinson<sup>1,2</sup>, Gregory Schill<sup>2</sup>, Troy D. Thornberry<sup>2</sup>, Patrick R. Veres<sup>5</sup>, Eleanor Waxman<sup>1,2</sup>, Andrew W. Rollins<sup>2</sup>

<sup>1</sup>Cooperative Institute for Research in Environmental Sciences, University of Colorado, Boulder, CO, USA

<sup>2</sup>NOAA Earth System Research Laboratory, Chemical Sciences Laboratory, Boulder, CO, USA

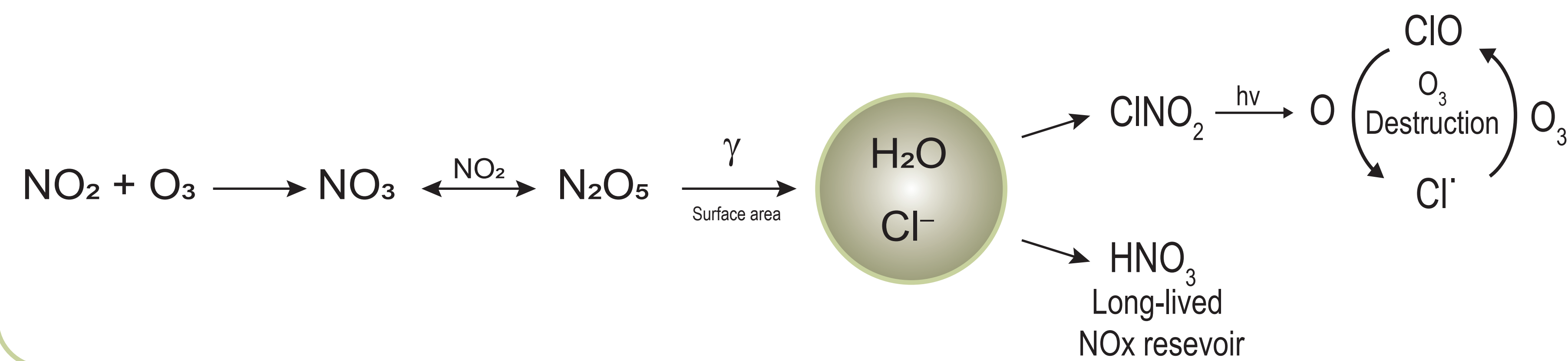
<sup>3</sup>NASA Langley Research Center, MS 483, Hampton, VA, USA

<sup>4</sup>Atmospheric Chemistry and Dynamics Branch, NASA Ames Research Center, Moffett Field, CA, USA

<sup>5</sup>National Center for Atmospheric Research, Boulder, CO, USA

## Reactive Nitrogen Influences Stratospheric $O_3$

The formation and fate of  $N_2O_5$  influences the total reactive nitrogen ( $NO_y$ ) budget. Heterogeneous (gas / particle) processing of  $N_2O_5$  transfers Nitrogen Oxides ( $NO_x$ ) into  $HNO_3$  (a long-lived  $NO_x$  reservoir) or nitryl chloride ( $ClNO_2$ , a source of  $Cl$  radicals) which in turn influence stratospheric ozone ( $O_3$ ) destruction. Yet, the  $N_2O_5$  uptake coefficient ( $\gamma$ ) of stratospheric aerosol has yet to be directly constrained by stratospheric observations and understanding of this parameter is instead based on laboratory and tropospheric field studies. As such, there is uncertainty in global climate models with respect to  $N_2O_5$  influences and how future changes in stratospheric aerosol might couple to  $O_3$  chemistry.



## An Observationally Constrained Iterative Diel Box Model Derives $\gamma(N_2O_5)$

- The model has been used to derive  $\gamma(N_2O_5)$  in the troposphere and lower-stratosphere
- The model begins 24 hours before the observation and iterates on initial conditions until the model and observations converge.
- Reactions (75 total) from the Master Chemical Mechanism are used including  $NO_x$ ,  $HO_x$ , and  $Cl_y$  reactions relevant to the stratosphere.

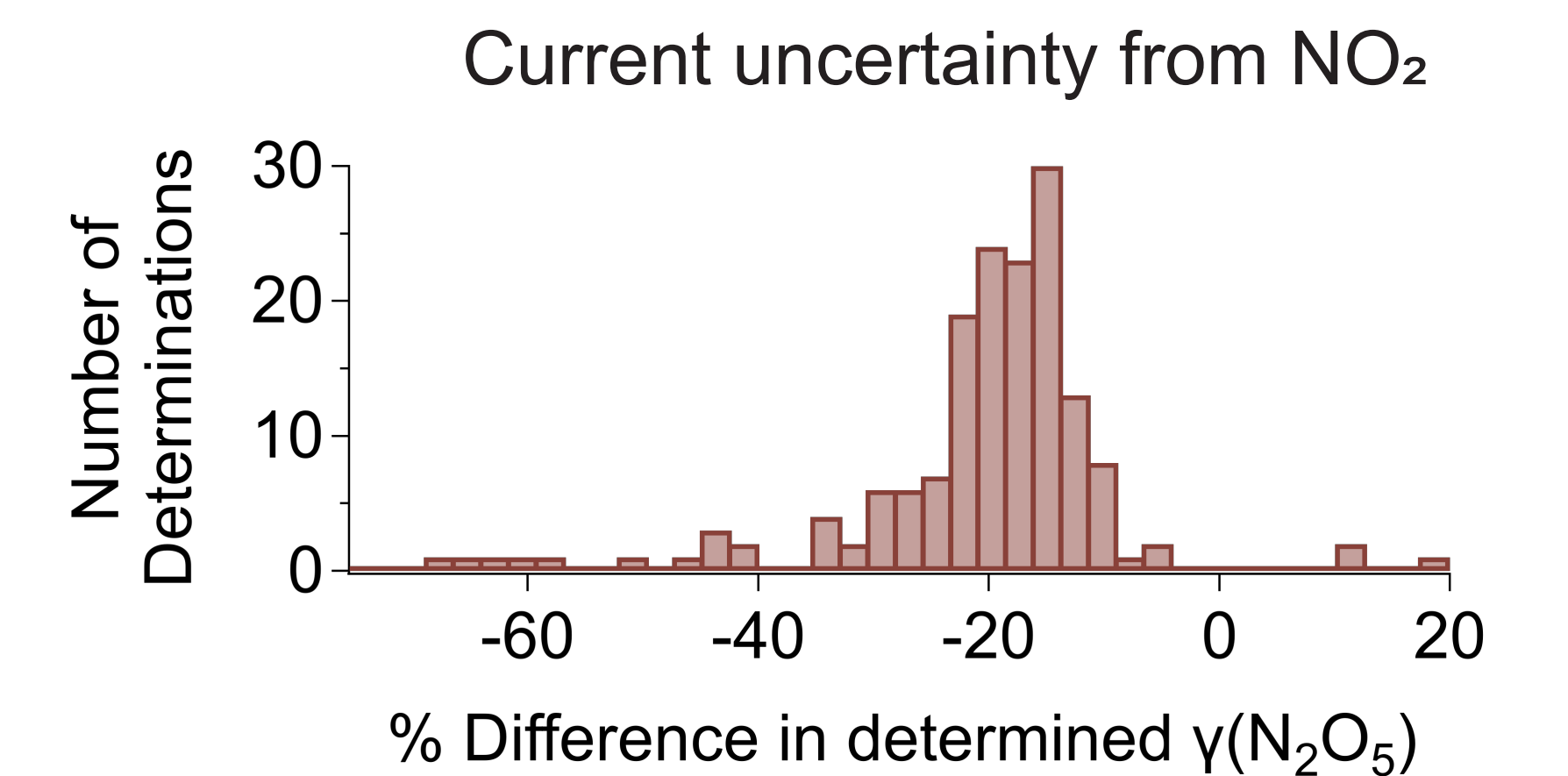
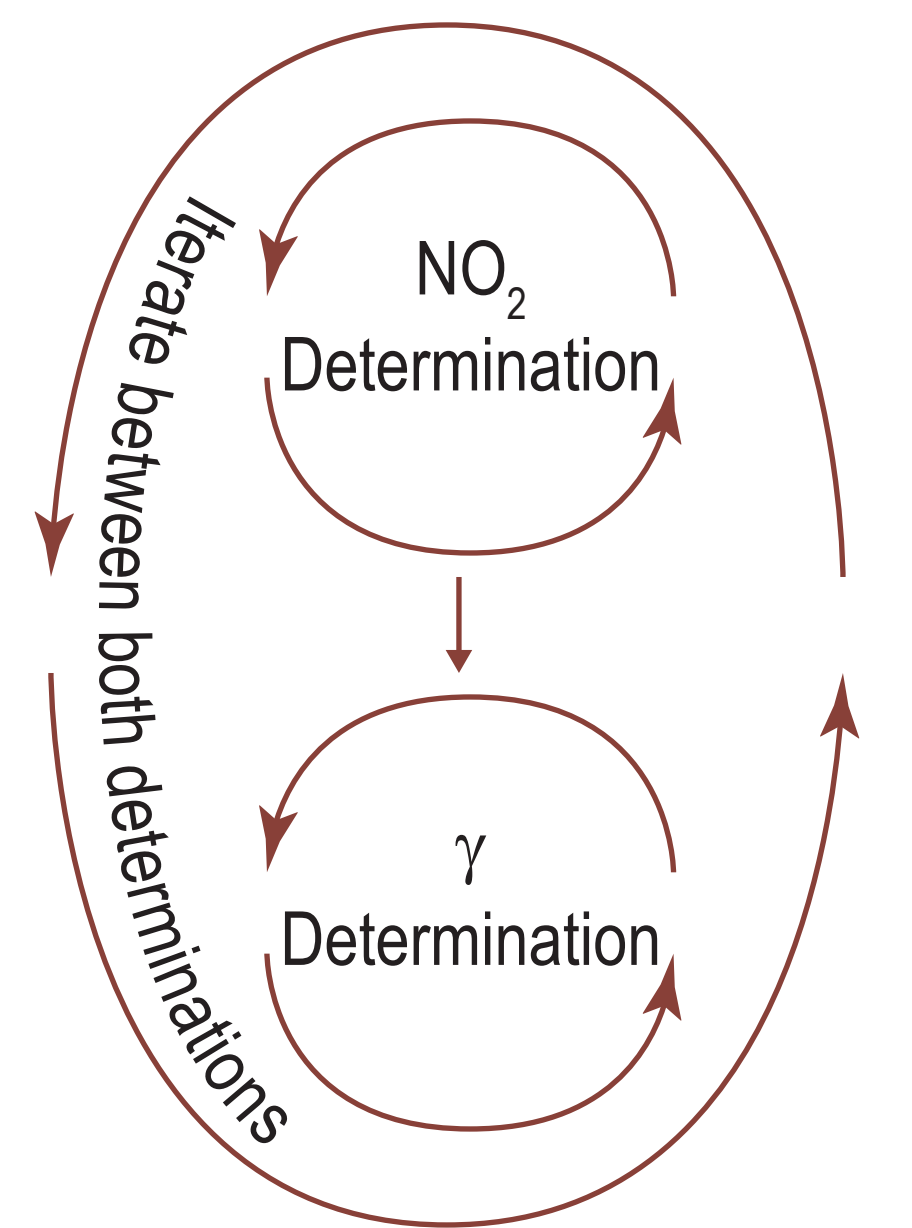
- For daytime observations,  $NO_2$  can be determined from observed total  $NO_x$ . The model iterates  $NO_2$  until modeled and observed  $NO$  are within 1%.
- $\gamma(N_2O_5)$  is determined by iterating  $\gamma$  until modeled and observed  $N_2O_5$  match within 2%.
- Both determinations are repeated until the derived  $\gamma$  remains within 5%.

### Model inputs include

- |                        |                             |
|------------------------|-----------------------------|
| Chemical               | Meteorology                 |
| • $N_2O_5$             | • Lat / Lon / Alt           |
| • $NO$                 | • 16 photolysis rates (TUV) |
| • $NO_2$               | • Temperature               |
| • $HNO_3$              | • Pressure                  |
| • $CO$                 |                             |
| • $O_3$                |                             |
| • Aerosol surface area |                             |

Using the modeled  $NO_2$  or observed  $NO_2$  results in a 20% difference in the  $\gamma(N_2O_5)$  result.

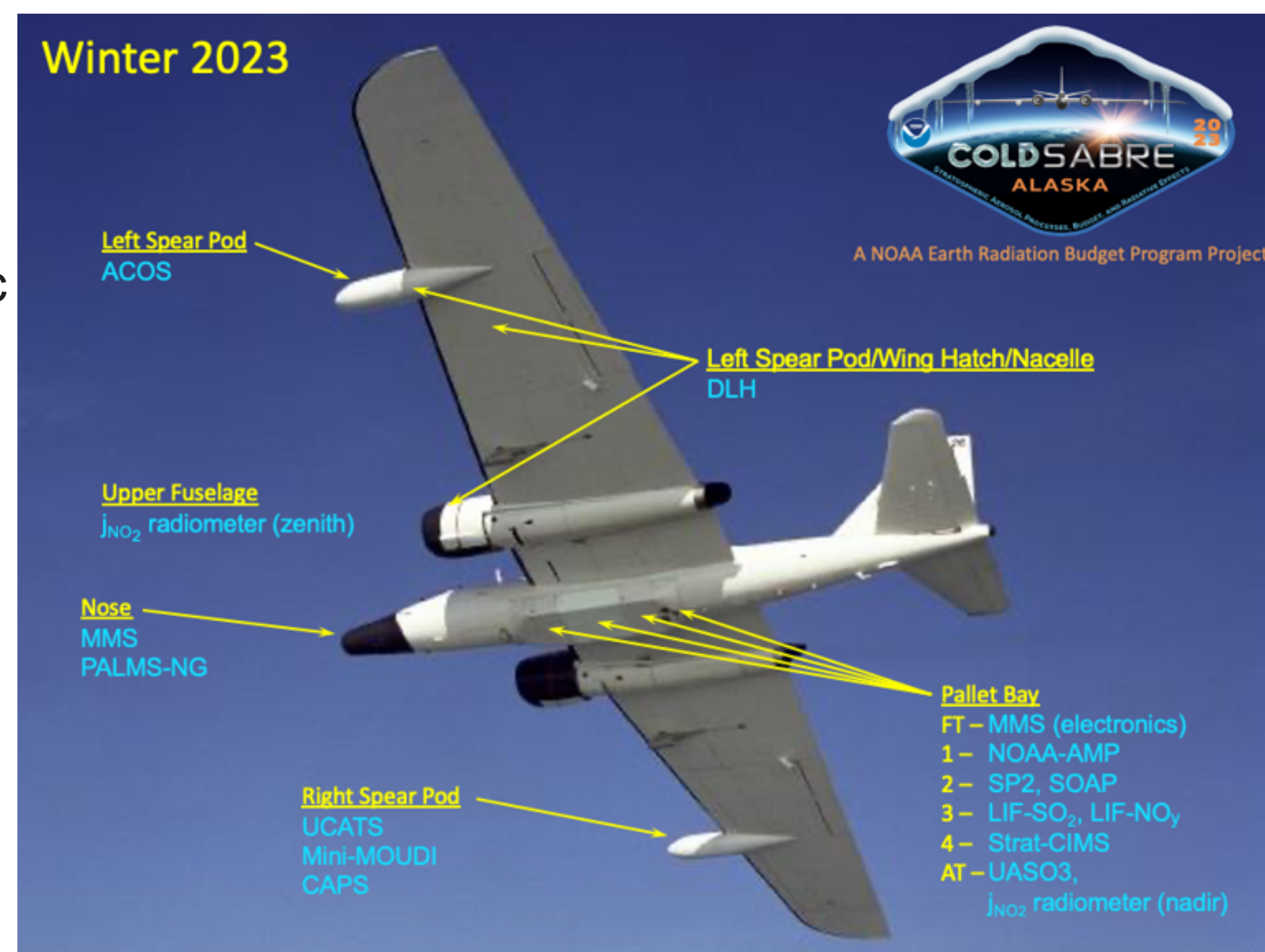
SABRE data is not yet final and these differences are being investigated. Data presented below use observed  $NO_2$ .



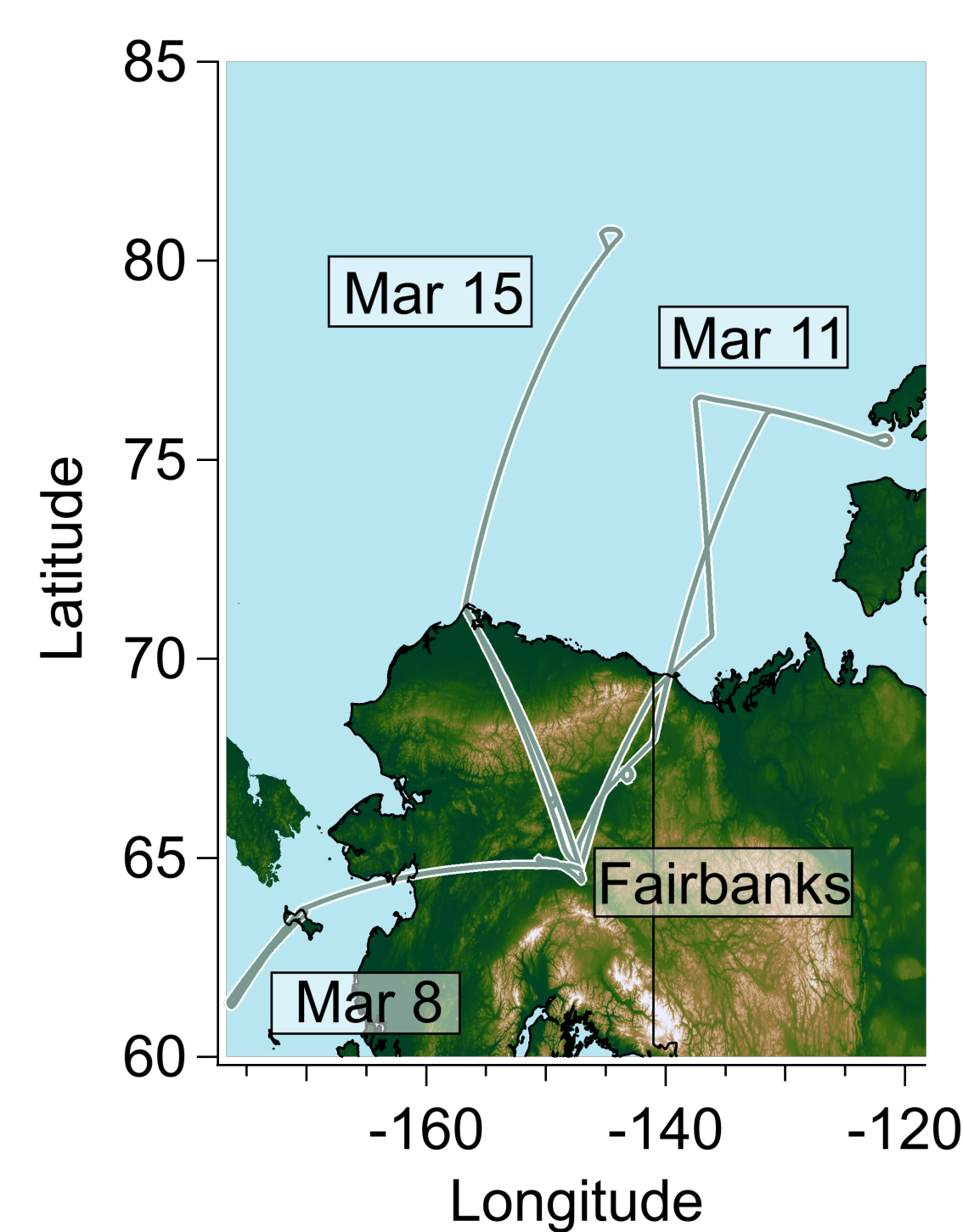
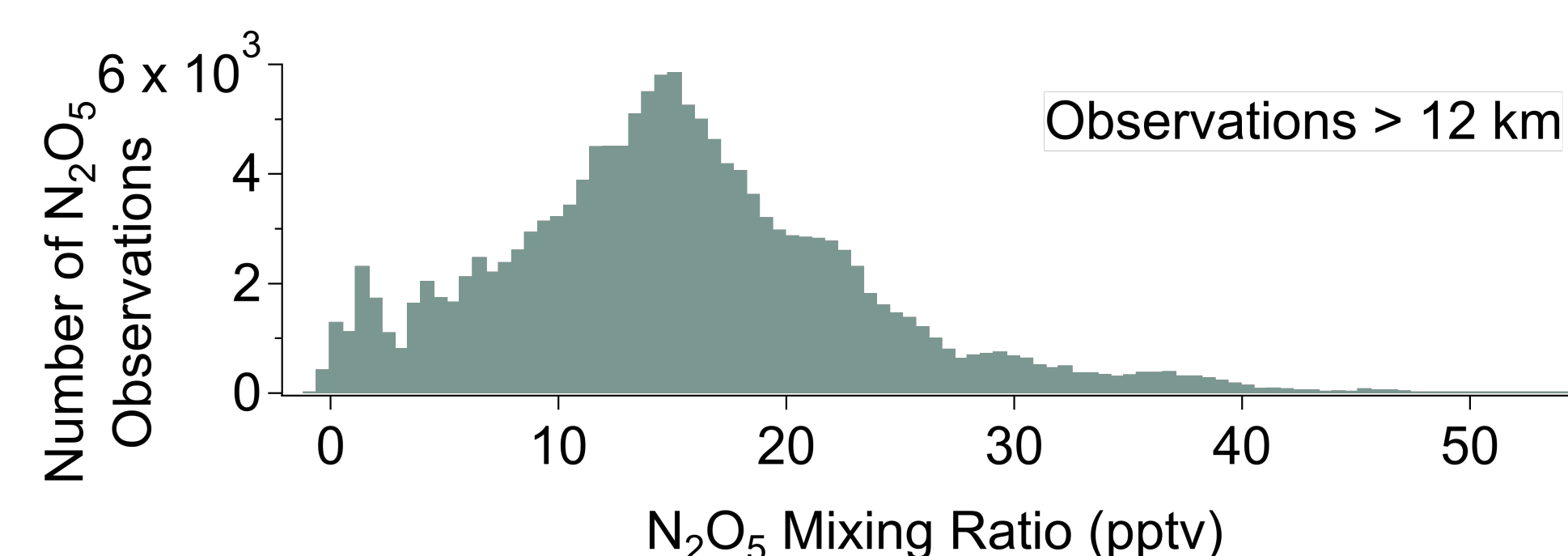
## The SABRE Alaska Campaign

In 2023 the NOAA Stratospheric Aerosol Processes, Budget, and Radiative Effects (SABRE) campaign achieved comprehensive observations of stratospheric gasses and aerosol from the NASA WB-57 high-altitude research aircraft.

- Includes 20 flights (12 at latitudes  $> 65^\circ$ )
- Altitudes range between 10 – 17 km
- Sampled in & out of winter polar vortex



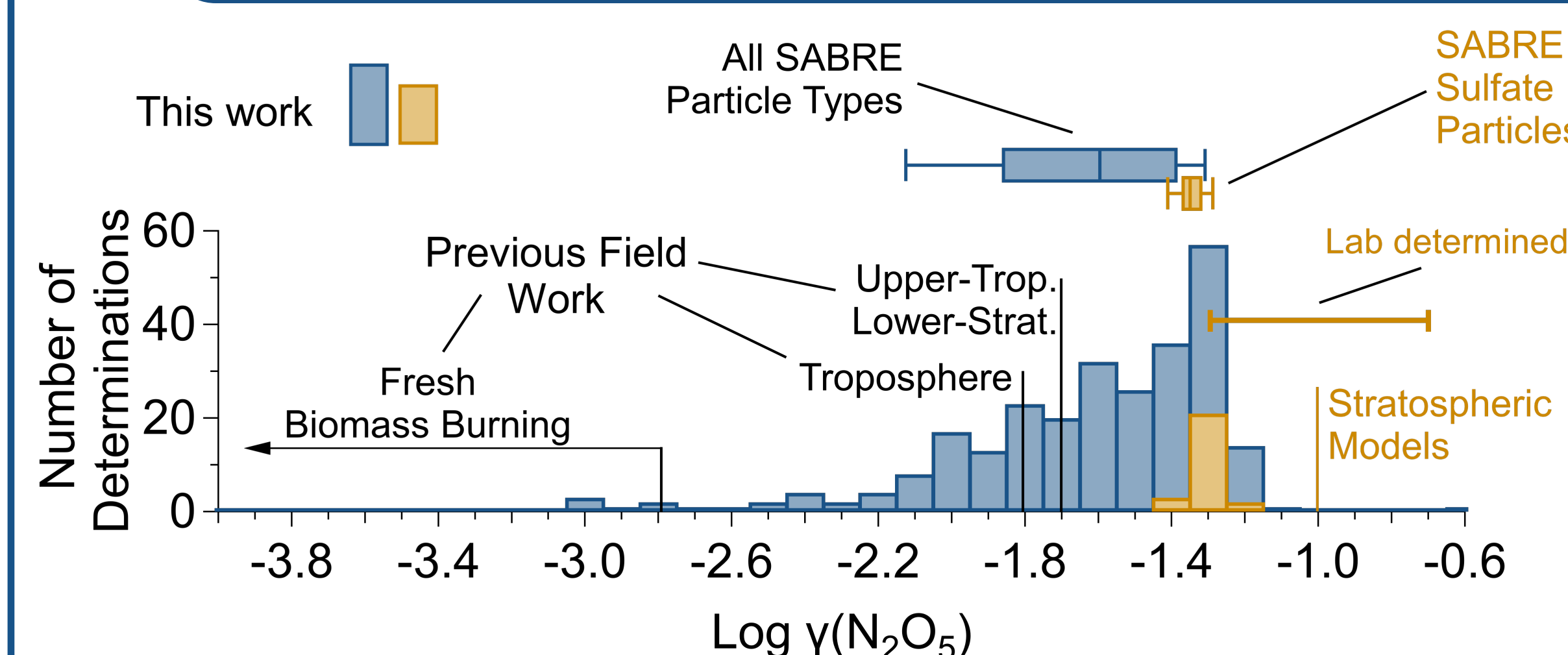
- The NOAA Stratospheric Chemical Ionization Mass Spectrometer (StratCIMS) achieved the first in-situ observations of  $N_2O_5$  which allows the determination of  $\gamma(N_2O_5)$ .



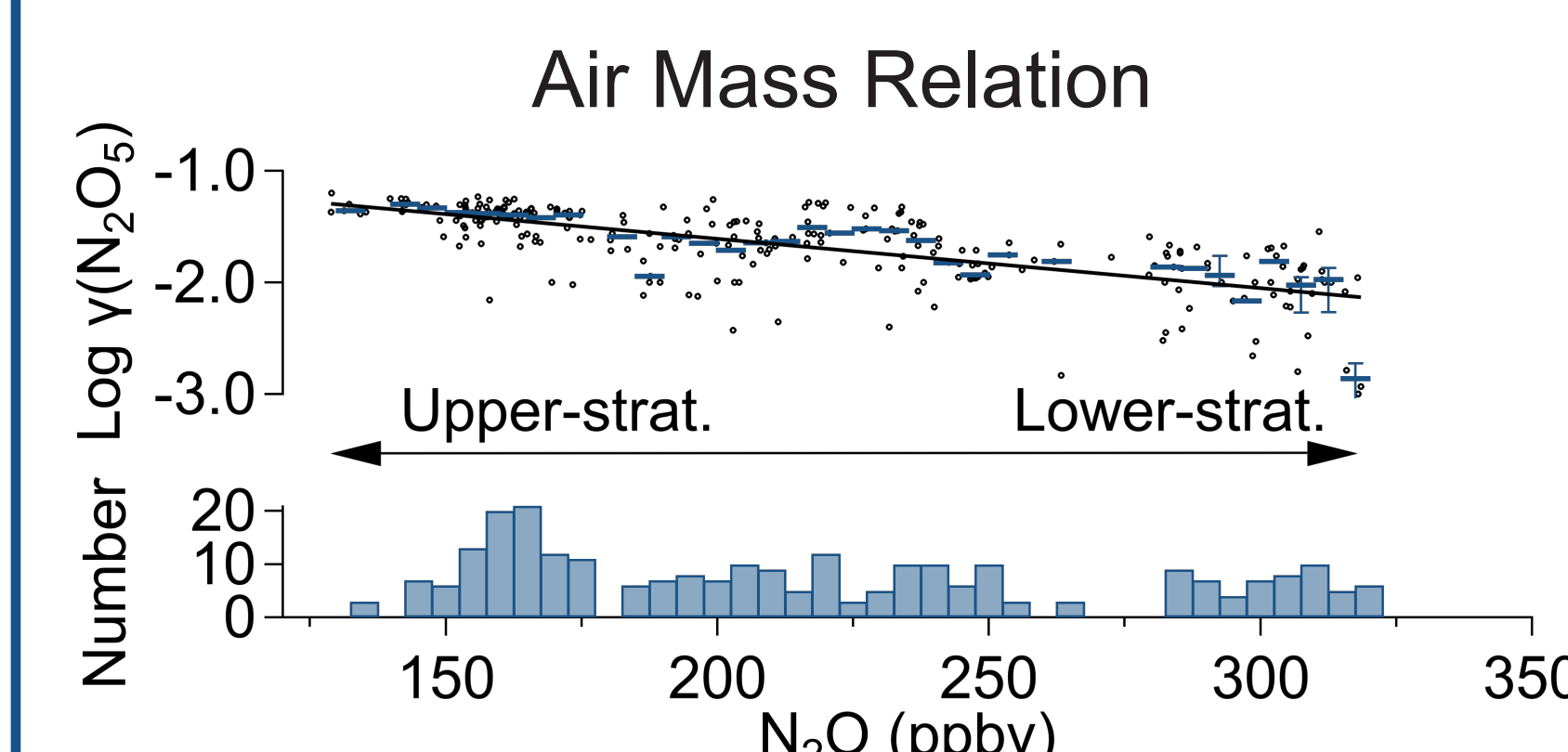
Our preliminary analysis uses SABRE data from three flights: Mar 8<sup>th</sup>, 11<sup>th</sup>, and 15<sup>th</sup>, which cover a variety of air mass types below

- Daytime and Nighttime
- Mid-latitude UTLS
- High-latitude non-vortex stratosphere
- Highly aged polar vortex

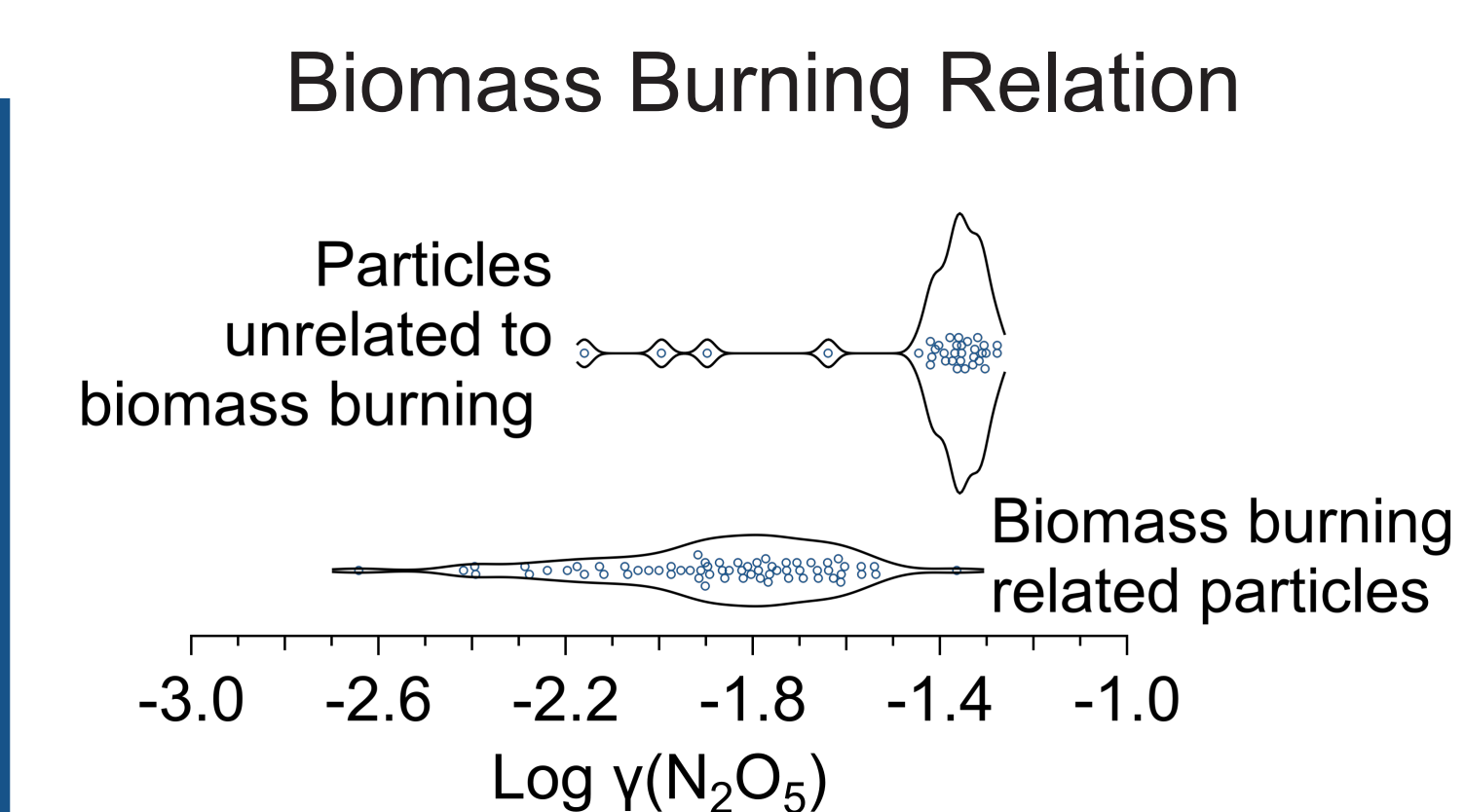
## Observationally Constrained Stratospheric $\gamma(N_2O_5)$ [Preliminary Results]



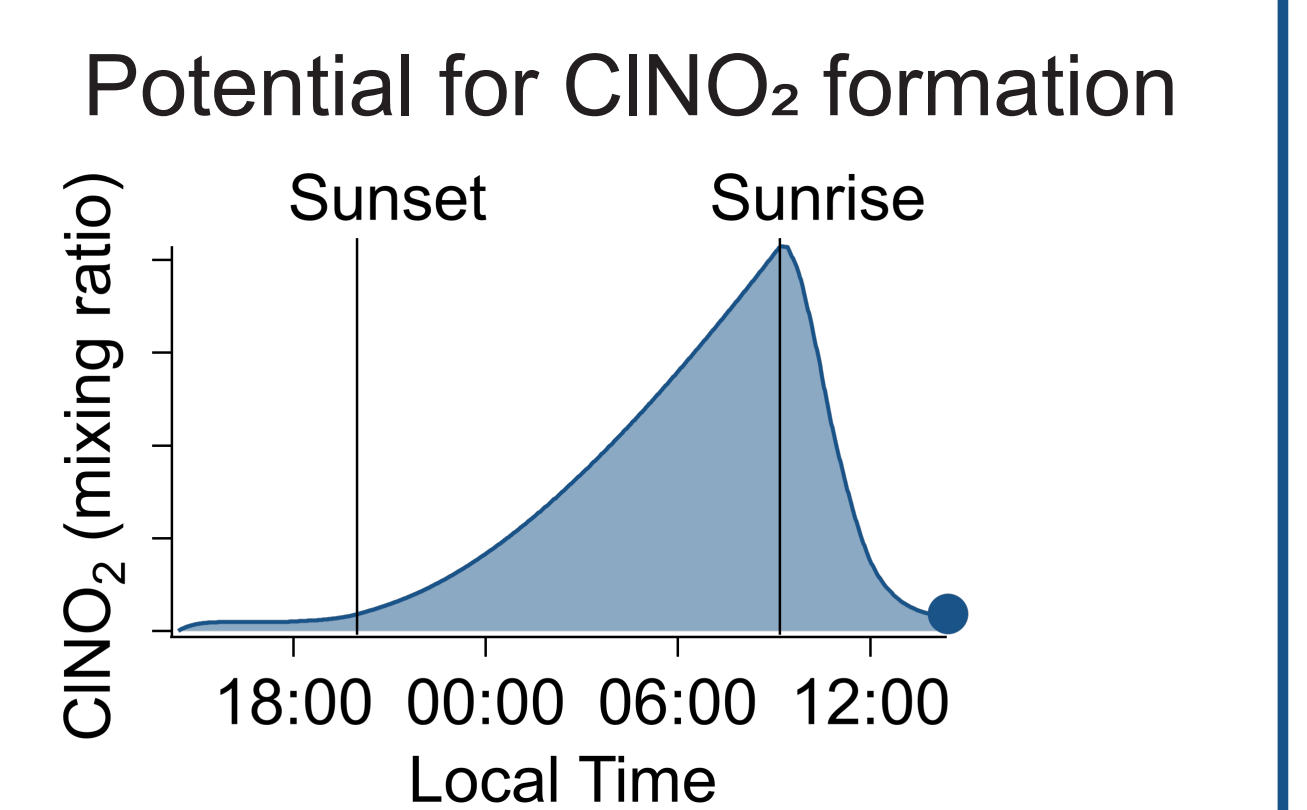
- Results suggest a median  $\text{Log } \gamma(N_2O_5)$  of  $-1.6$  (or  $2.5 \times 10^{-2}$ ) for all particle types.
- Results are consistent with previous field determinations of UTLS and Tropospheric derived values.<sup>1,2,3</sup>
- Stratospheric models assume pure sulfate particles and use a value roughly  $\times 2$  greater



The  $\gamma(N_2O_5)$  is correlated ( $R^2 \sim 0.44$ ) with air mass indicators such as  $N_2O$ . This relation is caused by changes in particle content such that  $\gamma(N_2O_5)$  increases as deeper regions of the stratosphere are sampled.



The  $\gamma(N_2O_5)$  on particles associated with biomass burning is consistently lower (a 33% median difference) consistent with field and laboratory studies.



Solomon et al. suggest organics increase  $HCl$  stratospheric particle solubility. With available particle chloride,  $ClNO_2$  formation is possible but currently unconfirmed.

## Next Steps

The implications of a lower  $\gamma(N_2O_5)$  in stratospheric models is being investigated. Specifically, what are the implications for  $O_3$ ? How sensitive are  $O_3$  outcomes to  $\gamma(N_2O_5)$  in a stratospheric aerosol injection geoengineering scenario? Further, observations of  $ClNO_2$  from the SABRE campaign will be used to determine if an upper-bound  $ClNO_2$  yield can be determined.

## References

- Novak et al. *in prep*
- Decker et al. *Geophys. Res. Lett.*, **2024**
- McDuffie et al. *JGR Atmos.*, **2018**
- Solomon et al. *Nature*, **2023**