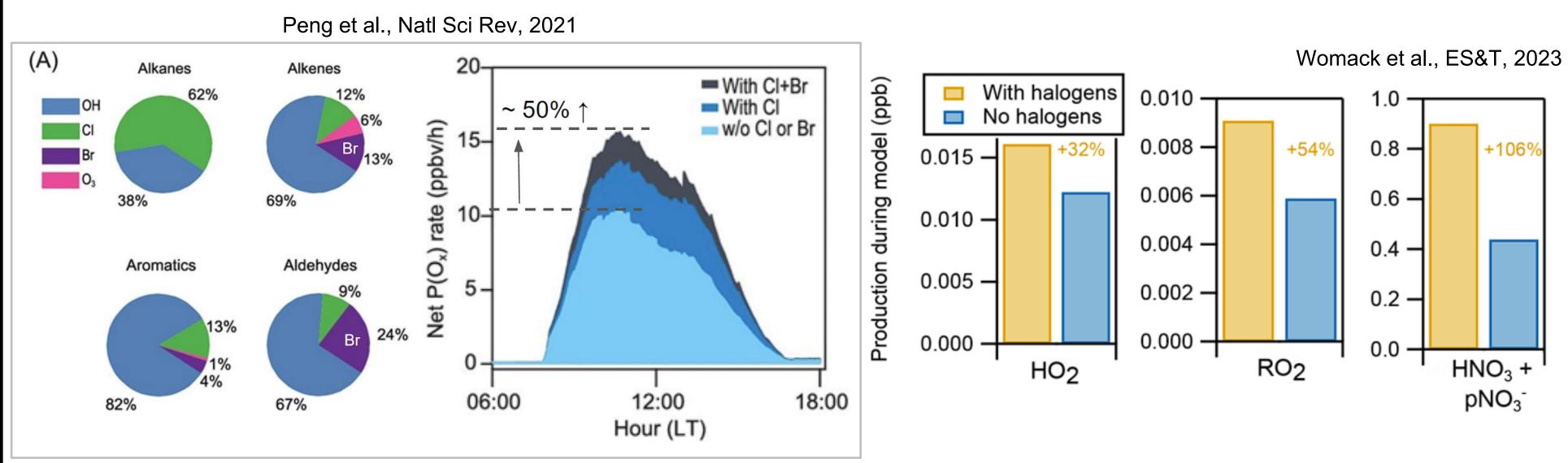
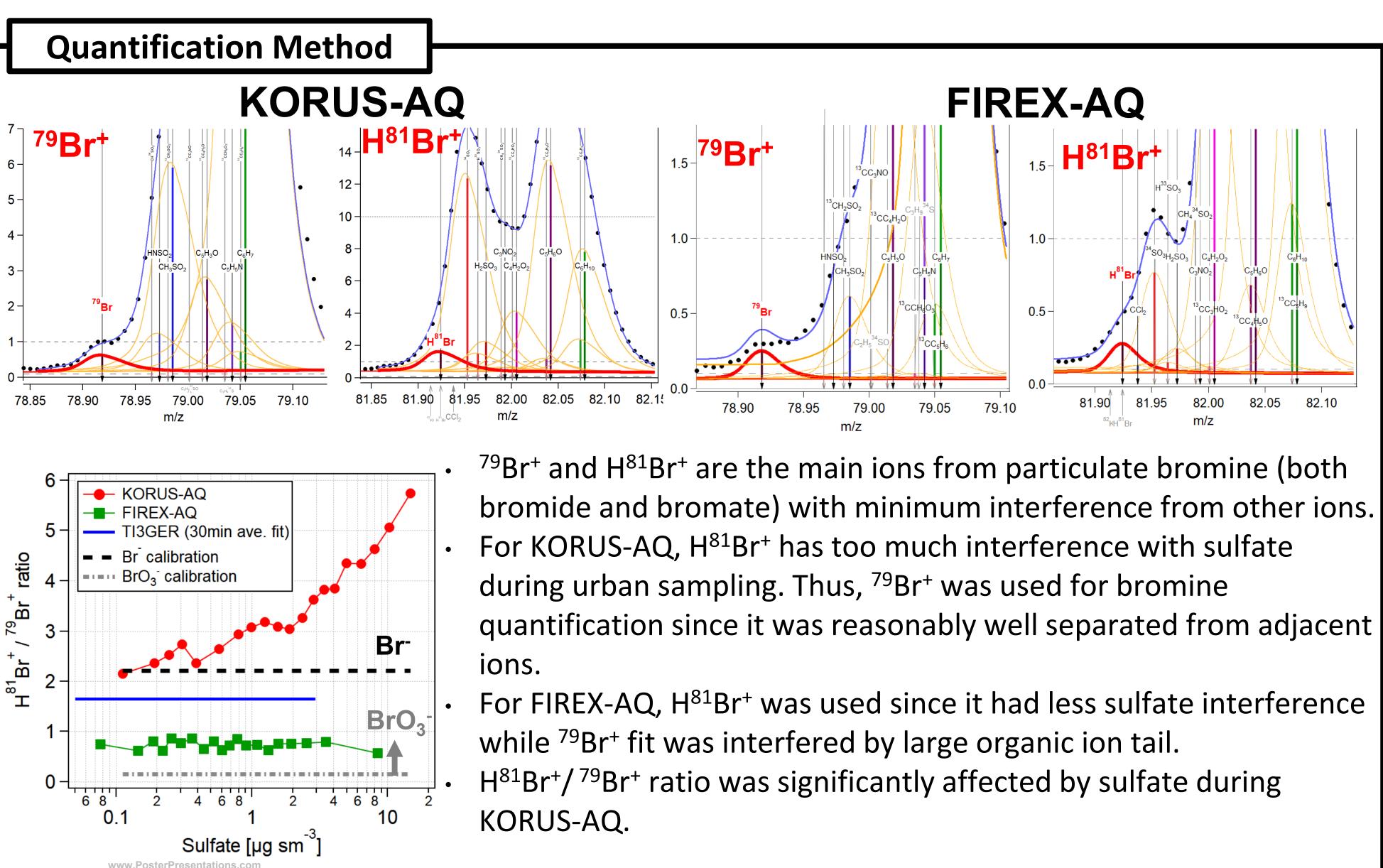


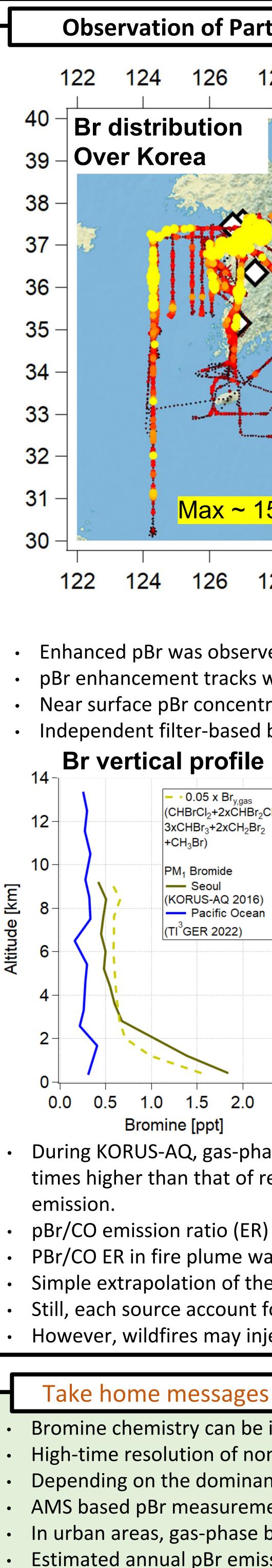
- Troposphere: ~ 30% of bromine is thought to be in particle phase (GEOS-Chem, Wang et al., 2021)
- Bromide is recycled to gas phase by various heterogeneous reactions and by photolysis of nitrate. Reactive Br₂ gas forms Br radical oxidizing VOCs in the atmosphere thereby enhancing O₃ and
- leading to secondary organic aerosol (SOA).
- HBr partitions preferentially into particle phase (pKa = -9.0), similar to sulfuric acid.

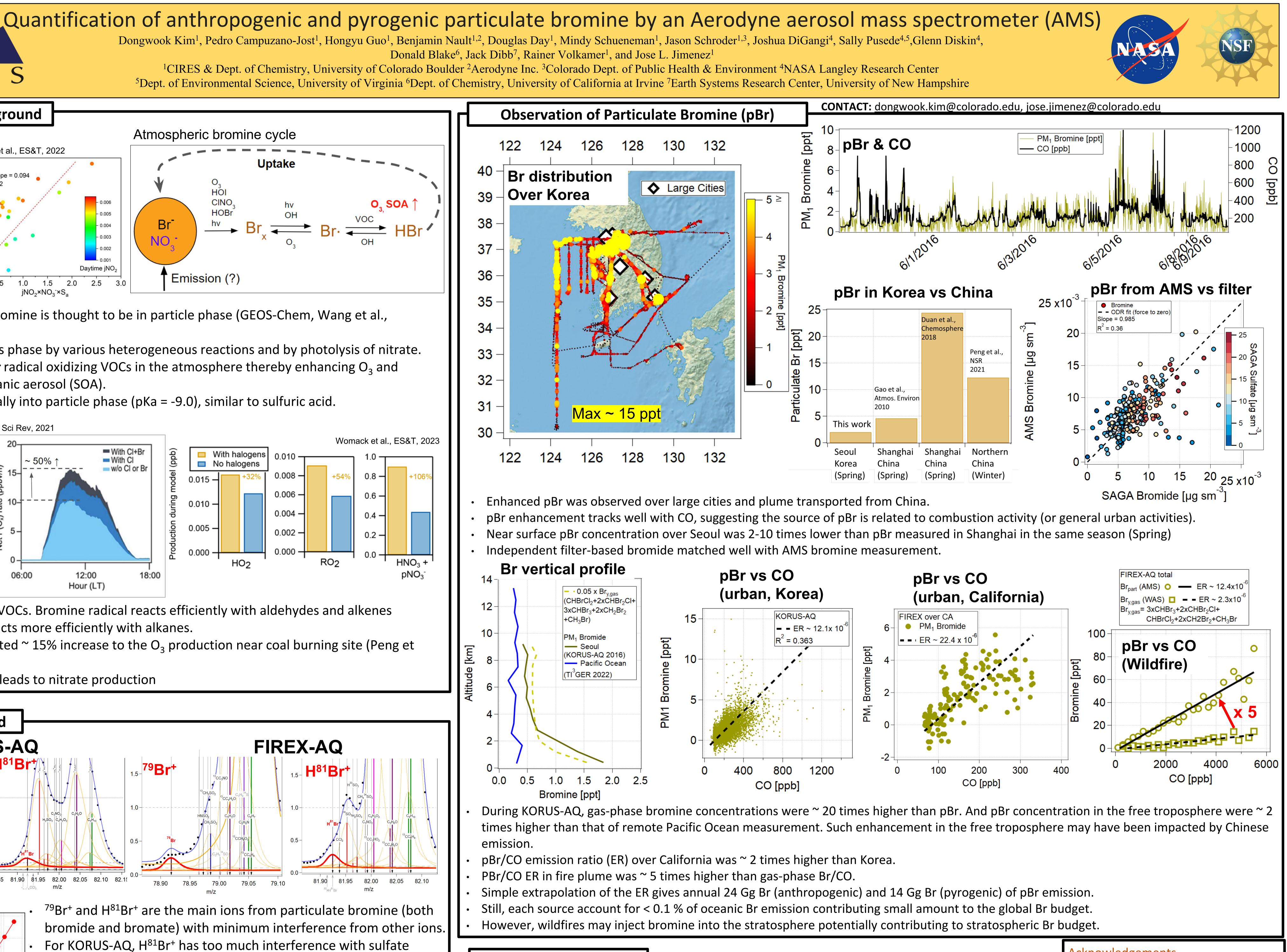


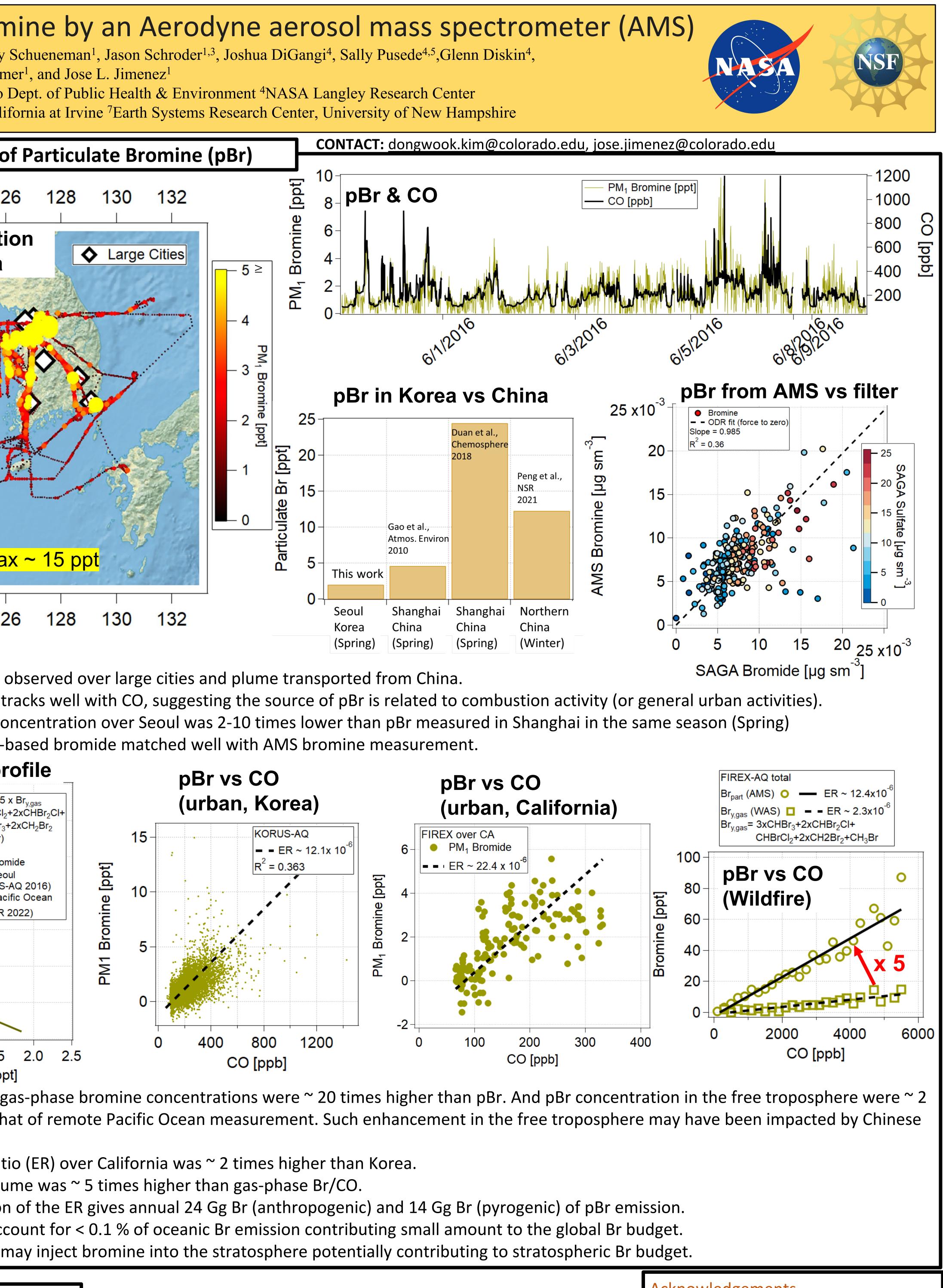
- Halogen radicals oxidize VOCs. Bromine radical reacts efficiently with aldehydes and alkenes while chlorine radical reacts more efficiently with alkanes. Bromine radical contributed ~ 15% increase to the O_3 production near coal burning site (Peng et
- al. 2021)
- Emission of bromine gas leads to nitrate production

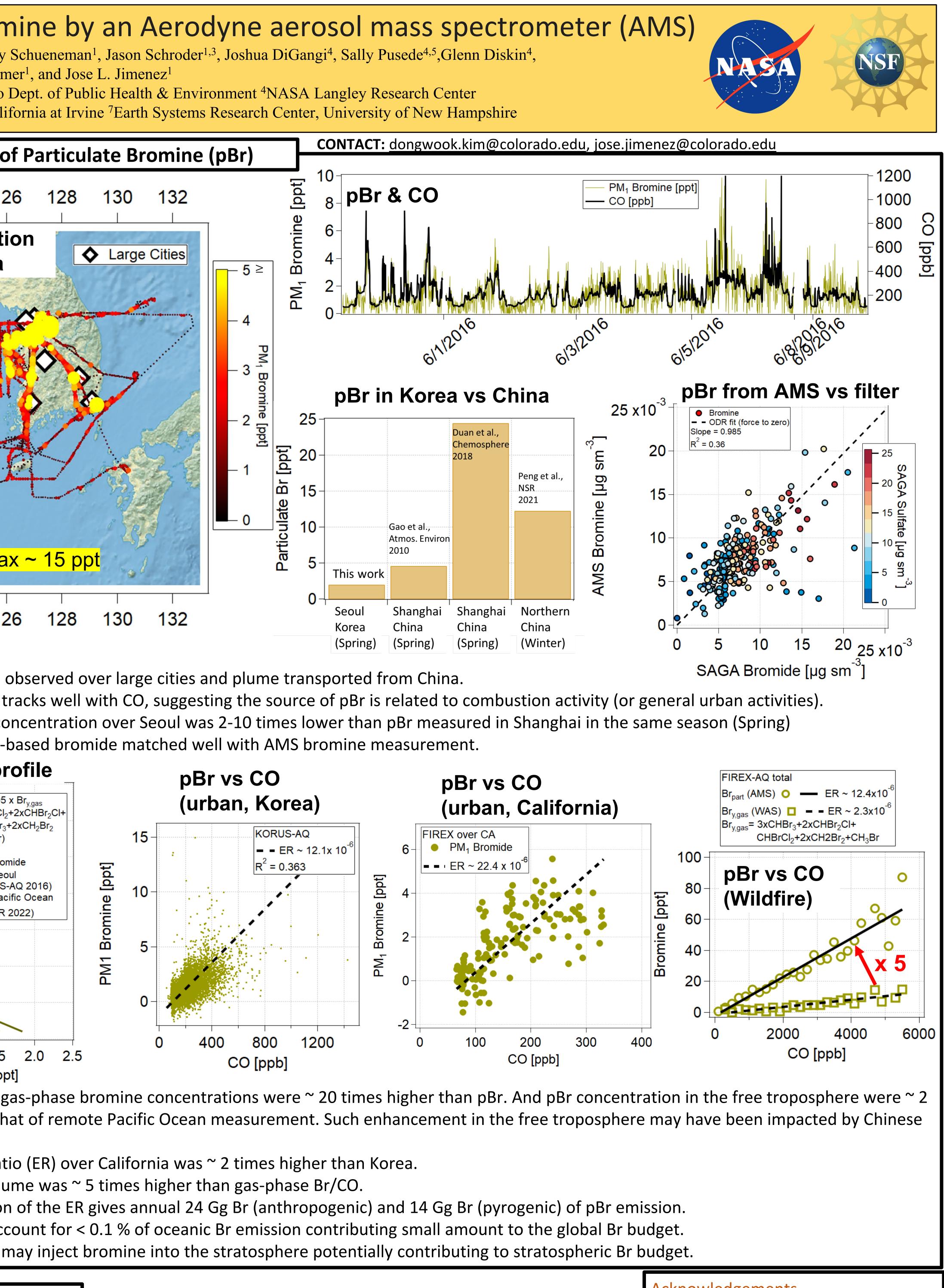


Donald Blake⁶, Jack Dibb⁷, Rainer Volkamer¹, and Jose L. Jimenez¹ ¹CIRES & Dept. of Chemistry, University of Colorado Boulder ²Aerodyne Inc. ³Colorado Dept. of Public Health & Environment ⁴NASA Langley Research Center









PBr/CO ER in fire plume was ~ 5 times higher than gas-phase Br/CO.

Bromine chemistry can be important for regional aerosol and O_3 formation.

High-time resolution of non-refractory pBr measurement was demonstrated using AMS. Depending on the dominant ambient aerosols, different tracer ions may need to be used AMS based pBr measurement agreed well width filter-based bromide measurement ove In urban areas, gas-phase bromine is dominant. In fire plumes, particulate bromine is do Estimated annual pBr emission: 24 Gg Br (anthropogenic) and 14 Gg Br (pyrogenic) Injection of wildfire-Br into the stratosphere merits further investigation.

	Acknowledgements
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