The Role of H₂S in Archean **Organic Haze Chemistry**

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INTRODUCTION

- > Methane-produced organic haze, CO_2 , and volcanic/biological H_2S were likely present in the Archean atmosphere.
- > Current view of Archean atmosphere separates their respective chemistries.
- > Organic haze production is currently thought to decrease with high CO $_2$ to CH $_4$ concentration ratios (greater than 1:1).

CURIOUSITIES EXPLORED

- \succ What is the interplay between H₂S and haze formation chemistry? How does this change as a function of CO2?
- > Could this interplay impact climate, habitability, or atmospheric chemistry of the Archean Earth?



METHODS

- Generate haze particles with trace amounts of H_2S via a flow system and UV reaction cell.
 - Analyze with quadrupole aerosol mass spectrometry (Q-AMS) or Scanning **Mobility Particle Sizer** (SMPS).
- > Measure the particle composition, size, and number in real time.



Laboratory experiments showed trace H₂S increased organic aerosol production in high-CO₂ organic haze chemistry. Signs of the formation of organosulfur and sulfate aerosol were also observed.





With 5 ppmv H₂S, the total aerosol mass loading increases with increasing initial CO₂ concentration



Standard sulfate fragmentation patterns could imply a mixture of sulfate and organosulfate aerosol



Particle density is higher with H_2S and initially increases with CO_2 , but slightly decreases after 0.5% CO_2

CONCLUSIONS & SIGNIFICANCE

- \succ Trace amounts of H₂S (5 ppmv) in Archean-like gas mixtures produced organic and sulfate aerosol, even at higher CO_2 : CH_4 concertation ratios.
- > There was no S_8 found at each CO₂ concentration, and only low H_2SO_4 at higher CO_2 concentrations.
- **Evidence leaned toward** organosulfur and organosulfate aerosol.
- > These results *defer from the* current thought of Archean atmospheric sulfur reservoirs.
- > This may be significant for interpreting the sulfur isotopic records, understanding the **Archean atmosphere/climate** and the early evolution of the atmosphere.



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